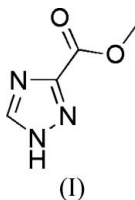


Methyl 1*H*-1,2,4-triazole-3-carboxylateXiang-Hai Guo^{a*} and
Qing-Xian Wang^b^aCollege of Pharmaceuticals & Biotechnology,
Tianjin University, Tianjin 300072, People's
Republic of China, and ^bTianjin Wuqing
Beiyang Chemicals Factory, Tianjin 301715,
People's Republic of ChinaCorrespondence e-mail:
kwoxhfly@yahoo.com.cn

Key indicators

Single-crystal X-ray study
T = 294 K
Mean $\sigma(\text{C}-\text{C}) = 0.002 \text{ \AA}$
R factor = 0.037
wR factor = 0.111
Data-to-parameter ratio = 13.4For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.The title compound, C₄H₅N₃O₂, possesses a planar triazole ring and contains two intermolecular hydrogen bonds in the crystal structure.

Comment

Ribavirin (1-*D*-ribofuranosyl-1,2,4-triazole-3-carboxamide) (Vo *et al.*, 2003) is a nucleoside analogue that has demonstrated efficacy in treating viral diseases both as monotherapy 14 (respiratory syncytial virus) and in combination therapy with interferon alpha (hepatitis C virus). Methyl 1*H*-1,2,4-triazole-3-carboxylate, (I) (Lin & Liu, 1984), has been used as a starting material for ribavirin (Ramasamy *et al.*, 2000). The structure of (I) (Fig. 1) displays two types of intermolecular hydrogen-bonding interactions, O1...H1—C1 and N1...H3—N3. The plane-to-plane distance of two molecules is 3.26 (2) Å. Molecules lying in the same plane are centrosymmetric and molecules in different planes are related by mirror symmetry.

Experimental

5-Amino-1,2,4-triazole-3-carboxylic acid (100 g) and methanol (500 ml) were placed in a 2 l three-necked flask with mechanical stirring. To the reaction mixture was slowly added 98% sulfuric acid (250 g) with stirring, and the mixture was heated under reflux for 16 h. The reaction mixture was cooled to 278 K for 10 h to afford a light-yellow wet solid. This solid was mixed with 98% sulfuric acid (58 g) and water (350 ml), and the resulting mixture cooled to 273–274 K. 30% aqueous sodium nitrite (150 g) was added slowly and the reaction allowed to continue for a further 2 h to give a grey solid. The grey solid and methanol (350 ml) were then placed in a 1 l flask with stirring and slowly heated to 313 K. When all the diazonium salt had been decomposed by methanol, the reaction solution was filtered and the filtrate was cooled to 283 K to afford the product, (I). Recrystallization from water and methanol gave 49 g of (I) (yield 49%).

Crystal data

C₄H₅N₃O₂
M_r = 127.11
Monoclinic, *P*2₁/*n*
a = 3.9737 (9) Å
b = 18.160 (4) Å
c = 8.1865 (19) Å
 β = 102.596 (4)°
V = 576.5 (2) Å³
Z = 4*D_x* = 1.464 Mg m⁻³
Mo *K*α radiation
Cell parameters from 1220
reflections
 θ = 2.2–26.2°
 μ = 0.12 mm⁻¹
T = 294 (2) K
Block, colourless
0.30 × 0.26 × 0.20 mm

Data collection

Bruker SMART CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.960$, $T_{\max} = 0.976$
 3192 measured reflections

1168 independent reflections
 846 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.023$
 $\theta_{\text{max}} = 26.4^\circ$
 $h = -2 \rightarrow 4$
 $k = -22 \rightarrow 21$
 $l = -10 \rightarrow 10$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.037$
 $wR(F^2) = 0.111$
 $S = 1.04$
 1168 reflections
 87 parameters
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0566P)^2 + 0.0868P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.002$
 $\Delta\rho_{\text{max}} = 0.15 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.16 \text{ e } \text{\AA}^{-3}$

Table 1

Selected geometric parameters (\AA , $^\circ$).

O1—C3	1.2008 (19)	N3—C1	1.318 (2)
N1—C1	1.323 (2)	N3—H3	0.95 (3)
N1—C2	1.353 (2)	C1—H1	0.9300
N2—C2	1.318 (2)	C4—H4C	0.9600
N2—N3	1.347 (2)		
C1—N3—H3	130.6 (15)	N1—C1—H1	124.8
N2—N3—H3	119.0 (15)	N1—C2—C3	121.13 (14)
N3—C1—H1	124.8	O2—C4—H4A	109.5
C2—N2—N3—C1	0.12 (18)	C1—N1—C2—C3	-179.58 (15)
C2—N1—C1—N3	0.14 (19)	N1—C2—C3—O1	3.1 (3)
N3—N2—C2—N1	-0.03 (18)	N1—C2—C3—O2	-176.59 (14)

Table 2

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C1—H1 \cdots O1 ⁱ	0.93	2.57	3.138 (2)	120
N3—H3 \cdots N1 ⁱ	0.95 (3)	1.88 (3)	2.822 (2)	175 (2)

Symmetry codes: (i) $x - \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$.

C-bound H atoms were positioned geometrically and refined in the riding-model approximation, with C—H = 0.93 and 0.96 \AA and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The H atom bound to N3 was refined with the distance restraint N—H = 0.95 (3) \AA .

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

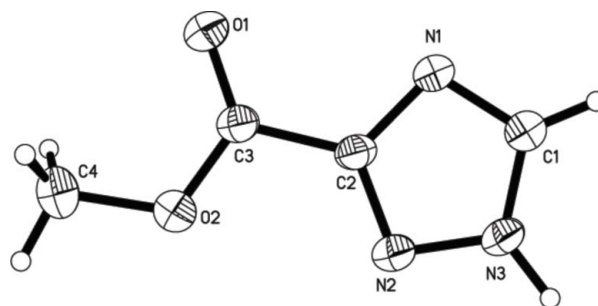


Figure 1

A view of the molecule of (I). Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii.

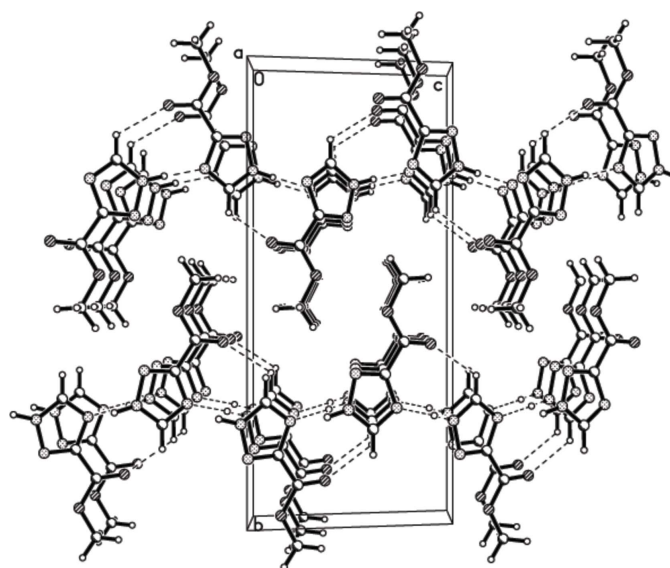


Figure 2

A packing diagram of (I). Dashed lines represent intermolecular hydrogen-bonding interactions.

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